

## [(1-Phenylethyl)iminomethyl]ferrocene

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## Key indicators

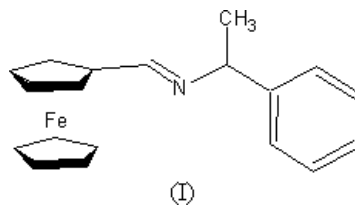
Single-crystal X-ray study  
*T* = 213 K  
 Mean  $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$   
*R* factor = 0.028  
*wR* factor = 0.084  
 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title Schiff base compound,  $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_{14}\text{H}_{14}\text{N})]$ , consists of discrete monomeric molecules separated by van der Waals contacts. The  $\text{C}=\text{N}$  bond length is 1.267 (3) Å. The ferrocenyl unit and the alkyl moiety bonded to the N atom of the imino group are arranged in *trans* positions.

## Comment

During the past decade optically active ferrocenyl compounds containing N-, O- and P-donor atoms have attracted considerable attention due to their possible application in the catalysis of enantioselective reactions (Richards & Locke, 1998; Togni, 2000). The structures of several Schiff base complexes containing a ferrocene unit have been reported (Carboni & Monnier, 1999; Enders *et al.*, 1997). The  $\text{C}=\text{N}$  bond distance [1.267 (3) Å] in the imino group of the title compound, (I), is generally consistent with those reported for other complexes of this type. The alkyl moiety in (I) occupies a *trans* position with respect to the ferrocenyl unit. The five membered rings of the ferrocene fragments are planar and parallel [dihedral angle = 3.2 (6)°].



## Experimental

The title compound, (I), was prepared by reaction of 1-phenylethylamine (20 mmol) with ferrocenealdehyde (20 mmol) in tetrahydrofuran. The reaction mixture was refluxed for 2 h and the solvent removed *in vacuo* to give an orange precipitate. The solid was collected and dried *in vacuo*. Suitable crystals were obtained by cooling a saturated solution of (I) in diethyl ether. Analysis calculated for  $\text{C}_{19}\text{H}_{19}\text{FeN}$ : C 71.92, H 6.00, Fe 17.66%; found: C 72.11, H 6.23, Fe 17.38%.

## Crystal data

$[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_{14}\text{H}_{14}\text{N})]$   
 $M_r = 317.20$   
 Monoclinic,  $P2_1$   
 $a = 5.9480$  (3) Å  
 $b = 7.5010$  (4) Å  
 $c = 17.3197$  (10) Å  
 $\beta = 92.751$  (1)°  
 $V = 771.84$  (7) Å<sup>3</sup>  
 $Z = 2$

$D_x = 1.365 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation  
 Cell parameters from 3511 reflections  
 $\theta = 2-25^\circ$   
 $\mu = 0.97 \text{ mm}^{-1}$   
 $T = 213$  (2) K  
 Block, orange  
 $0.40 \times 0.30 \times 0.20 \text{ mm}$

Data collection

Bruker SMART CCD  
diffractometer  
 $\omega$  scans  
Absorption correction: multi-scan  
(SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.698$ ,  $T_{\max} = 0.830$   
5063 measured reflections

3511 independent reflections  
3264 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$   
 $\theta_{\text{max}} = 28.8^\circ$   
 $h = -7 \rightarrow 6$   
 $k = -9 \rightarrow 10$   
 $l = -23 \rightarrow 19$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.028$   
 $wR(F^2) = 0.084$   
 $S = 1.12$   
3511 reflections  
266 parameters  
All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.056P)^2 + 0.0168P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.007$   
 $\Delta\rho_{\text{max}} = 0.25 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.26 \text{ e } \text{Å}^{-3}$   
Absolute structure: Flack (1983),  
1416 Friedel pairs  
Flack parameter = 0.011 (16)

Table 1

Selected geometric parameters (Å, °).

C14—C12	1.514 (3)	N1—C12	1.482 (3)
N1—C11	1.267 (3)	C12—C13	1.527 (4)
C11—N1—C12	115.5 (2)	N1—C12—C13	108.3 (2)
N1—C12—C14	110.7 (2)	C14—C12—C13	112.1 (2)

All H atoms were located in a difference Fourier map and freely refined.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics:

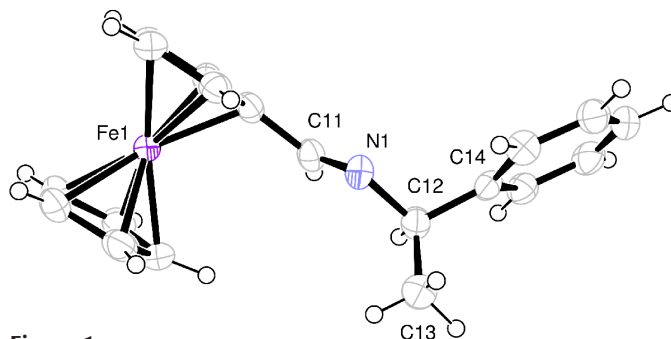


Figure 1  
View of the molecule of (I) showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.

ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2003).

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References

Bruker (2000). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.  
Carboni, B. & Monnier, L. (1999). *Tetrahedron*, **55**, 1197–1201.  
Enders, D., Peters, R., Lochtmann, R. & Runsink, J. (1997). *Synlett*. pp. 1462–1464.  
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.  
Richards, C. J. & Locke, A. J. (1998). *Tetrahedron Asymmetry*, **9**, 2377–2407.  
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.  
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.  
Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.  
Togni, A. (2000). *J. Organomet. Chem.* **598**, 365–368.